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# DEGRADATION AND FORMATION OF SULFOLIPID OCCURRING CONCURRENTLY WITH DE- AND RE-GENERATION OF CHLOROPLASTS IN THE CELLS OF CHLORELLA PROTOTHECOIDES<sup>1</sup>

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1. It has been demonstrated that when the cells of Chlorella protothecoides are grown mixotrophically under illumination in a medium rich in nitrogen source (urea) and poor in glucose, the normal green cells are obtained, while in a medium rich in glucose and poor in the nitrogen source, entirely chlorophyll-less cells with profoundly degenerated plastids ("glucose-bleached" cells) are produced, irrespective of whether in the light or in darkness. The "glucose-bleached" cells turn green with regeneration of fully organized chloroplasts when incubated in a nitrogen-enriched medium in the light ("light-greening"), while in the dark they become pale green with formation of only partially organized chloroplasts ("dark-greening"). When, on the other hand, the green cells are transferred into a medium enriched with glucose, they are bleached fairly rapidly with degeneration of chloroplasts in the light as well as in darkness ("bleaching"). Using <sup>35</sup>S as a tracer, investigations were made on the changes of contents of the algal cells in sulfolipid and other sulfur compounds during the processes of the greening and bleaching.

2. By determining the radioactivities of chromatographically separated sulfur-containing compounds of the uniformly  ${}^{35}$ S-labeled green ("G") and "glucose-bleached" ("W") cells, it was found that the concentration of a species of sulfolipid (discovered by BENSON et al.) as well as those of glutathione, sulfotrioses and most of the other sulfurcontaining compounds were at least 5 times higher in the "G" cells than in the "W" cells, while sulfoquinovosyl glycerol was present in approximately equal amounts in the two types of cells.

3. Phospholipid contents and compositions in the two types of algal cells were found to be practically identical.

4. The sulfolipid content of algal cells increased and decreased almost in parallel with the processes of greening and bleaching, respectively.

5. Studying the mode of incorporation of radiosulfate into various sulfur compounds of algal cells during the processes of "light- and

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dark-greening" and "bleaching" (lasting about 70 hr), it was found that active <sup>35</sup>S-incorporation into sulfolipid occurred throughout the process of "light-greening," while in the "dark-greening" and "bleaching" the active incorporation abruptly ceased after the initial 24 hr period of experiments. It was suggested that the biosynthesis of the sulfolipid is closely related to the formation of photosynthetic apparatus in chloroplast.

6. When the <sup>35</sup>S-labeled green cells were bleached in a medium containing no radiosulfate, the <sup>35</sup>S-sulfolipid and most of other <sup>35</sup>S-sulfur compounds decreased markedly but the <sup>35</sup>S-sulfoquinovosyl glycerol increased considerably. It was inferred that the deacylation of the sulfolipid, a surfactant lipid, with formation of water soluble sulfoquinovosyl glycerol may be a cardinal event of bleaching process, causing a disintegration of the intact architechture of photosynthetic apparatus.

7. Based on these observations it was concluded that the sulfolipid is an integral component of photosynthetic structure.

It has been known that a sulfolipid is contained in various photosynthetic organisms including algae, bacteria and green leaves (1-4), and it was revealed to be 6-sulfo- $\alpha$ -D-quinovopyranosyl- $(1\rightarrow 1')$ -2', 3'-di-O-acyl-D-glycerol (2-9). In all photosynthetic plants thus far examined its content was found to equal or exceed those of any of the phospholipids (2, 4, 10). BENSON et al. (4, 10) assumed that this sulfolipid forms—by virtue of its surfactant nature—an important architectural component in the oriented matrix of pigments, enzymes, etc. in the photosynthetic apparatus.

SHIHIRA-ISHIKAWA and HASE (11) have recently demonstrated the reversible degeneration and regeneration of chloroplasts ("bleaching" and "greening") in the cells of *Chlorella protothecoides* by controlling light conditions as well as the concentration balance between glucose (or other organic carbon sources) and nitrogen source. Taking advantage of this characteristic of the alga, investigations were made of the problem as to whether and in what manner the contents of the solfolipid and other sulfur-containing compounds would change during the processes of the de- and re-generation of chloroplasts. The observations to be described in the present paper seem to provide strong experimental evidence in favor of the assumption made by BENSON et al.

#### MATERIAL AND METHODS

The strain of *Chlorella protothecoides* and the culture (shaking) apparatus used were the same as those reported by Shihira-Ishikawa and Hase (11). The temperature and light intensity used in the culture were  $23^{\circ}$  and 2 klux throughout the present experiments.

## Preparation of the "green" (G) and "glucose-bleached" (W) cells

The methods of preparation of these cells were the same as those de-

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scribed previously (12): the green cells were obtained by growing, in the light, the alga for 5 days in the medium containing 0.5% urea and 0.1% glucose in addition to the basal mineral nutrients of the following composition: per liter,  $\text{KH}_2\text{PO}_4$ , 0.7 g;  $\text{K}_2\text{HPO}_4$ , 0.3 g;  $\text{MgSO}_4 \cdot 7\text{H}_2\text{O}$ , 0.3 g;  $\text{FeSO}_4 \cdot 7\text{H}_2\text{O}$ , 3 mg;  $\text{A}_{\text{RNON}}$ 's " $\text{A}_5$ " solution (13) 1 ml; thiamine hydrochloride, 10 µg. The entirely chlorophyll-less "glucose-bleached" cells were obtained by culturing the alga in the medium containing, besides the basal nutrients, 0.1% urea and 1.0% glucose, under otherwise the same conditions as those used for the green cells.

# "Greening" and "bleaching" experiments

In the "greening" experiment the glucose-bleached cells separated from the preculture medium were incubated in the medium—referred to G-medium—containing 0.5% urea and the basal mineral nutrients but without addition of glucose, in the light ("light-greening") or in darkness ("darkgreening").

In the "bleaching" experiment the green cells were incubated, in darkness, in the medium containing 1.0% glucose and the basal mineral nutrients in the absence of urea. This medium shall be called B-medium in the following. In the control expriment the green cells were incubated in the basal mineral medium. During the 72 hr period of these incubations, aliquots of the cell suspensions were withdrawn at intervals from the cultures and subjected to the determinations of packed volume and chlorophyll content of cells (14), as well as to chromatographic analyses of the sulfolipid and related compounds.

#### Labeling of the cells with radioisotopes

Algal cells uniformly labeled with  ${}^{35}S$  were prepared by growing the cells in the above media containing radiosulfate (1 mc/mM). Starting with these  ${}^{35}S$ -labeled cells or with the non-labeled cells, the following eight sets of experiments were carried out.

- (A) Greening experiments
  - (I) Labeled W-cells were incubated in labeled G-medium under illumination.
  - (II) Non-labeled W-cells were incubated, under illumination, in labeled G-medium, to investigate the <sup>35</sup>S-incorporation during the process of "light-greening."
  - (III) The same as above, but without illumination, to investigate the <sup>35</sup>S-incorporation during the process of "dark-greening."
- (B) Bleaching experiments
  - (IV) Labeled G-cells were incubated in labeled B-medium.
  - (V) Non-labeled G-cells were incubated in labeled B-medium.
  - (VI) Labeled G-cells were incubated in non-labeled B-medium.
- (C) Control experiments

(VII) Non-labeled G-cells were incubated in labeled basal medium.

(VIII) Labeled G-cells were incubated in non-labeled basal medium.

Cells uniformly labeled with <sup>32</sup>P were also prepared for phospholipid analyses using the media containing inorganic phosphate-<sup>32</sup>P.

# Determination of algal components

For the determination of sulfur-containing or phosphor-containing compounds on a microscale, radiochemical procedures were extensively employed. Radioactivity of a compound from the uniformly labeled cells represented its relative molar concentration.

Algal suspensions, well washed with water, were directly applied to the origin of chromatogram and subjected immediately to chromatography. Quantitative extraction of all compounds were effected with the chromatographic solvent as development proceeded. The omission of previous extraction of the compounds enabled to avoid their possible fractionation and oxidation. Radioactivities of the spots separated on paper chromatograms were directly measured with a wide area, end-window G. M. tube (Nihon Musen Co., Tokyo, Type 5006), and corrected for disintegration per minute with the use of average counting efficiencies obtained with a liquid scintillation spectrometer.

Two types of chromatography were routinely employed: two-dimensional chromatography on Toyo No. 51 paper (Toyo Roshi Co., Tokyo) first with phenol-water (500:130 w/w), second with butanol-propionic acid-water (142:71:100 v/v/v) (15), and a descending ion-exchange resin-paper chromatography of lipid mixtures on Amberlite WB-2 (Rohm and Haas Co., Philadelphia) using diisobutylketone-acetic acid-water (8:5:1 v/v/v). The former chromatography separated most of the water-soluble components, and the latter complex lipid mixtures into more than 12 individual lipids.

Identification of the algal compounds was carried out either by cochromatography and co-electrophoresis on paper with authentic compounds or by chromatographic and electrophoretic investigation of their deacylation, hydrolysis or chemical modification products (10, 16).

# RESULTS AND DISCUSSION

# 1. Changes in chlorophyll content and volume of the algal cells during the processes of greening and bleaching

Characteristic changes in chlorophyll content and volume of the algal cells during the processes of greening and bleaching are shown in Fig. 1. When the "glucose-bleached" cells were incubated, under illumination, in the nitrogen-enriched mineral medium without added glucose, active synthesis of chlorophyll started after the lag period of about 15 hr, and continued for the following period of experiment ("light-greening"). In the dark, on the other hand, there occurred much less active formation of chlorophyll which soon came to a standstill ("dark-greening"). During these greening processes the cell volume increased slightly. When the

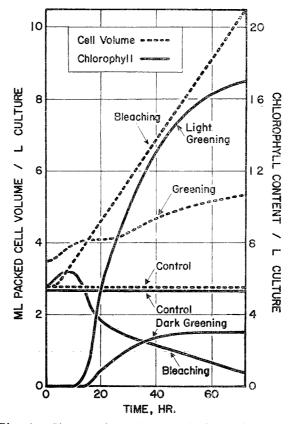


Fig. 1. Changes in volume and chlorophyll content of algal cells during the processes of greening and bleaching in the light and in darkness. For explanation, see text.

green cells were incubated in the glucose-enriched medium in the absence of the nitrogen source (in darkness), a marked degradation of chlorophyll took place after a slight increase during the initial short period, being accompanied by a considerable increase in the cell mass ("bleaching"). In the control culture, in which the green cells were incubated in the basal mineral medium without addition of glucose, no significant changes in chlorophyll and volume of the cells were observed.

Electronmicroscopic studies have revealed that during the processes of bleaching and greening a marked degeneration and regeneration, respectively, of chloroplast structures were occurring (11 and unpublished data). The cells obtained in the "light-greening" contain fully organized chloroplasts, while those in the "dark-greening" show only partially organized chloroplasts without lamellar structures. It has also been shown in separate studies that the formation and degradation of chlorophyll are, on the whole, paralleled by the appearance and disappearance, respectively, of photosynthetic activities of the algal cells, except that in the "dark-greening" no  $CO_2$ -fixing capacity of the cells is produced although their photo-phos-

phorylating power is developed (OH-HAMA, SHIHIRA-ISHIKAWA and HASE, to be published).

# 2. Differences in concentrations of sulfur-containing compounds between the green and "glucose-bleached" cells

As shown in Table I, the concentrations of the sulfolipid, an unidentified lipid<sup>2</sup> and glutathione were found to be markedly higher in the green cells than in the glucose-bleached cells. On the other hand, the deacylated sulfolipid, 6-sulfo- $\alpha$ -p-quinovopyranosyl- $(1\rightarrow 1')$ -glycerol, which was found in

### TABLE I

Concentrations of major sulfur-containing compounds in the green and glucosebleached cells of Chlorella protothecoides<sup>a</sup>

	<sup>35</sup> S cp	"G"/"W" ratio	
	"G" cells	"W" cells	"G / "W Talle
Sulfolipid	35,700	7,820	5
Sulfoquinovosyl glycerol	33,700	36,600	1
Glutathione	40,800	3,490	11
Unidentified lipid	32,100	425	76
Total soluble S-compounds	191,000	85,800	2

<sup>*a*</sup> Two types of algal cells uniformly labeled with <sup>35</sup>S (1 mc/mM) were analyzed by means of two-dimensional paper chromatography. The observed <sup>35</sup>S activities represent relative molar concentrations of individual components. Four compounds listed here constituted more than 90% of the algal sulfur content, excluding inorganic sulfate and insoluble macromolecules.

TABLE II

Phospholipid composition of the green (G) and glucose-bleached (W) cells of Chlorella protothecoides  $({}^{32}P-labeled)^a$ 

	"G"	cells	"W" c	ells	''G''/''W''	
	cpm <sup>a</sup>	%	cpm <sup>a</sup>	%		
Phosphatidyl choline	4,440	54.5	4,020	42.9	1.1	
Phosphatidyl ethanolamine	2,130	26.1	3,140	33.3	0.7	
Phosphatidyl glycerol	605	7.4	920	9.8	0.7	
Phosphatidyl serine	431	5.4	862	9.1	0.5	
Phosphatidyl inositol	540	6.6	484	5.1	1.4	
Total lipid-P	8,140	100	9,430	100	0.9	

<sup>a</sup> <sup>32</sup>P-Counts per minute per 0.1 ml of packed algal cells. Specific activity of radiophosphor was same in both types of cells, and the cpm represents relative molar concentration of each lipid.

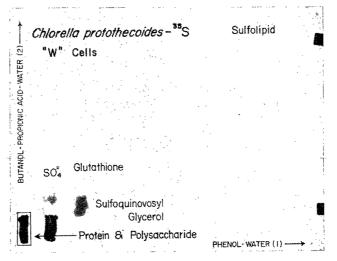
<sup>2</sup> Like in some other strains of green algae (16), a considerable <sup>35</sup>S-activity was found in an unknown spot obtained from "G"-cells on two-dimensional paper chromatograms. The spot moved as far as green pigments did, suggesting that the compound may be a lipid. This compound, whose structure analysis is in progress, is contained in extremely high concentrations in some plants such as *Lemna perpusilla* (SHIBUYA. MARUO and BENSON, in preparation).

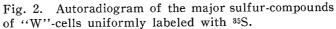
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the water-soluble fraction (16), was present in nearly the same concentrations in these two types of cells. In W-cells it formed the major sulfurcontaining compound. Experiments also revealed that the concentrations of most of the minor sulfur-containing components such as lysosulfolipid (17), methionine, taurine, cystine and sulfotrioses (16) were several times higher in "G"-cells than in "W"-cells.

#### 3. Phospholipids

Table II shows the results of analysis (Amberlite WB-2 chromatography) of phospholipids in <sup>32</sup>P-labeled "G" and "W" cells. In sharp contrast to





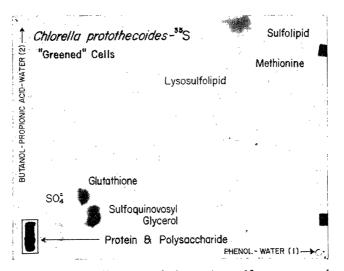


Fig. 3. Autoradiogram of the major sulfur-compounds of the 72 hours-greened cells uniformly labeled with <sup>35</sup>S.

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the results obtained with the sulfur-containing compounds, essentially the same composition of the major phospholipids was found with these two types of algal cells. A repeated analysis gave practically the same result.

4. Changes in quantities of sulfur compounds of algal cells during the processes of greening and bleaching

(*i*) Greening (Expt. A-I)—The glucose-bleached cells uniformly labeled with <sup>35</sup>S were incubated, in the light, in the greening medium containing radiosulfate of the same specific activity (1 mc/m) as that in the preculture, and the process of cell greening was followed for 72 hr. The specific activity of <sup>35</sup>S was, therefore, constant throughout the experiment, and the determination of the radioactivities of the sulfur-containing compounds separated from the algal cells served as a measure of their relative molar concentra-

# TABLE III

Change in relative quantities of the major sulfur-containing compounds of <sup>35</sup>S-labeled algal cells during the process of light-greening<sup>a</sup>

		Time (hr)									
		0	3	9	15	24	48	72			
Ptotein and polysaccharide <sup><math>b</math></sup>	$\frac{1}{2}$	4,010 7,110	5,710 9,280	7,120 13,000	9,570 19,600			,			
SO <sub>4</sub> =	$\frac{1}{2}$	7,610 13,500	12,800 20,800	13,000 23,700	$13,900 \\ 28,500$	25,700 50,800	25,100 77,300	16,000 60,200			
Sulfoquinovosyl glycerol	$\frac{1}{2}$	3,810 6,760	5,080 8,260	5,140 9,380	4,440 9,110	5,070 10,000	4,710 14,500	4,950 18,600			
Glutathione	$\frac{1}{2}$	292 518	335 545	472 862	677 1,390	2,710 5,350	2,490 7,670	3,610 13,600			
Sulfolipid	$\frac{1}{2}$	640 1,130	802 1,300	1,100 2,010	1,190 2,440	$1,640 \\ 3,240$	$3,790 \\ 11,700$	4,870 18,300			
Lysosulfolipid	$\frac{1}{2}$	57 101	85 138	$\begin{array}{c} 147\\ 318\end{array}$	197 404	475 938	354 1,090	571 2,150			
Methionine	$\frac{1}{2}$	89 158	$\begin{array}{c} 172 \\ 280 \end{array}$	338 617	387 793	283 558	406 1,250	1,250 4,700			
Total <sup>¢</sup>	$\frac{3}{4}$	16,500 29,300	24,700 40,600	27,300 49,900	30,400 67,200	55,400 109,000	58,900 181,000	52,700 198,000			

<sup>a</sup> The algal cell suspensions were withdrawn at intervals as indicated, and analyzed by two-dimentional paper chromatography. The radioactivity of <sup>35</sup>S in this table represents relative molar concentration of each compound since the specific activity was constant throughout the 72 hours' period of experiment.

1 : <sup>35</sup>S dpm per  $\mu$  liter of packed algal cells.

2: <sup>35</sup>S dpm per ml of culture.

3 :  $^{35}\mathrm{S}$  cpm per  $\mu$  liter of packed algal cells.

4 : <sup>35</sup>S cpm per ml of culture.

<sup>b</sup> Radioactivity remained at the origin of the chromatogram. Repeated extraction of this area did not cause loss of the activity.

<sup>c</sup> Radioactivity of the cell sample was measured on planchet without subjecting to chromatography.

tions in the cells. Figures 2 and 3 are the typical autoradiograms of the <sup>35</sup>S-compounds in the starting "W"-cells and the 72 hours-greened cells, respectively.

Table III shows the result of the determination of the radioactivities of sulfur-containing compounds in the cells at various stages of the lightgreening. Figure 4 illustrates the changes in concentration of the major sulfur-containing compounds in the cells during the process of light-greening. The sulfolipid increased gradually and its intracellular concentration became approximately 5 times higher after 72 hours' incubation.

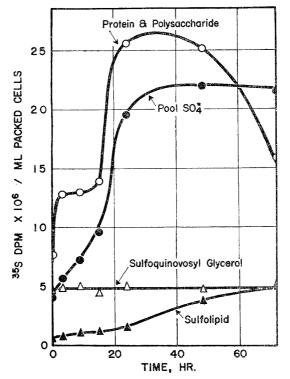


Fig. 4. Changes in content of major sulfurcontaining compounds of algal cells during the process of light-greening. Specific activity of <sup>35</sup>S was constant throughout the 72 hours' experiment.

Other minor compounds, such as sulfotrioses, behaved in almost the same manners as the sulfolipid. Radioactivities of the protein and polysaccharide fraction as well as of inorganic sulfate pool increased most rapidly after 15 hours and soon reached the maximum. Sulfoquinovosyl glycerol was the only compound whose content in cells remained almost constant throughout the experiment. It should be noted that the greening medium contained only urea besides the basal mineral nutrients, and the growth of the algal cells did not take place markedly because of the lack of exogenous carbon and energy sources. The cells at the end of the

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greening (designated as "greened" cells in Fig. 3), therefore, may be different from the "G"-cells, and the disagreement between the data obtained with the "greened" cells (Fig. 4 and Table III) and those with the "G"cells (Table II) is understandable.

(*ii*) Bleaching (Expt. B-IV)—A similar experiment at a constant  ${}^{35}$ S-specific activity was conducted for the process of bleaching. Table IV shows the radioactivities of the major sulfur compounds separated on two-dimensional chromatograms of the uniformly  ${}^{35}$ S-labeled cells withdrawn at intervals during the process of bleaching. It is seen from the table that the amount of sulfolipid (expressed per unit volume of culture) increased during the initial 24 hr period and decreased in the later period of experiment, while its content in cells (expressed per unit volume of packed cells) was almost constant during the initial 24 hr period but afterwards dropped sharply. As will be seen from the results of later experiments, these changes in

ΤА	BL	Æ	IV	

Changes in relative quantities of the major sulfur-containing compounds of <sup>35</sup>S-labeled algal cells during the process of bleaching<sup>a</sup>

	Time (hr)									
		0	3	9	15	24	48	72		
Protein and polysaccharide <sup>b</sup>		19,800 35,600	19,700 36,100	20,200 40,400	16,700 40,600	$14,600 \\ 44,500$	7,700 38,200	6,560 37,700		
SO4=	$\frac{1}{2}$	$1,140 \\ 2,050$	3,500 6,400	4,150 8,300	2,940 7,140	3,920 12,000	7,700 38,200	9,250 53,200		
Sulfoquinovosyl glycerol	$\frac{1}{2}$	2,900 5,220	$1,860 \\ 3,400$	2,880 5,760	3,180 7,730	4,650 14,200	5,610 27,800	6,790 39,100		
Glutathione	$\frac{1}{2}$		1,690 3,090	$1,140 \\ 2,280$	790 1,920	152 1,200	95 472	78 449		
Sulfolipid	$\frac{1}{2}$	,	2,160 3,950	1,910 3,820	2,350 5,710	2,460 7,510	1,080 5,400	945 5,440		
Lysosulfolipid	$\frac{1}{2}$	397 715	594 1,090	426 852	351 853	339 1,040	199 986	192 1,110		
Total <sup>o</sup>	3 4		1,490 2,730	1,460 2,920	1,210 2,940	1,090 3,320	727 3,600	763 4,390		

<sup>a</sup> The algal cell suspensions were withdrawn at intervals as indicated, and analyzed by two-dimensional paper chromatography. The specific activity of <sup>35</sup>S was constant throughout the 72 hours' transition, and the radioactivity in this table represents relative molar concentration of each compound.

1 : <sup>35</sup>S dpm per  $\mu$  liter of packed algal cells.

2 : <sup>35</sup>S dpm per ml of culture.

3 : <sup>35</sup>S cpm per  $\mu$  liter of packed cells.

4 : <sup>35</sup>S cpm der ml of culture.

 $^{b}$  Radioactivity remained at the origin of the chromatogram. Repeated extraction of this area did not cause any loss of the activity.

<sup>e</sup> Radioactivity of the cell suspension was directly measured on planchet without chromatography. Correction for counting efficiency was not made.

#### TABLE V

Incorporation of radiosulfate into various algal sulfur compounds during the process of light-greening<sup>a</sup>

		Time (hr)								
		$0^b$	3	9	15	24	48	72		
Protein and polysaccharide	$\frac{1}{2}$	25 108	143 677	$1,370 \\ 6,970$	7,250 31,000	12,600 65,900	$13,100 \\ 80,200$	14,600 92,700		
SO <sub>4</sub> =	$\frac{1}{2}$	80 352	779 3,710	2,520 12,900	6,880 35,200	4,520 23,600	7,570 46,800	9,290 59,000		
Glutathione	$1 \\ 2$	0 0	7 31	73 371	409 2,100	749 4,920	$1,420 \\ 8,750$	1,160 7,350		
Sulfoquinovosyl glycerol	$\frac{1}{2}$	0 0	61 288	204 1,080	1,700 8,650	1,720 9,030	5,030 31,000	6,160 38,800		
Sulfolipid	$\frac{1}{2}$	0 0	$\begin{array}{c} 10 \\ 47 \end{array}$	$\begin{array}{c} 82\\ 417\end{array}$	293 1,490	861 4,500	$2,450 \\ 15,100$	3,150 19,800		
Lysosulfolipid	$rac{1}{2}$	0 0	19 89	111 568	303 1,540	427 2,230	828 5,080	753 4,760		

<sup>a</sup> Non-labeled algal cells were incubated in an <sup>35</sup>S-radioactive medium and the aliquotes were withdrawn at intervals as indicated. The values in this table show the radioactivity incorporated into the chromatographically separated sulfur compounds. For further explanation see the legend of Table III and text.

<sup>b</sup> Intracellular radioactivities found at zero time are due to the incorporation during a very short period of time elapsed between the inoculation and withdrawal of the aliquot.

#### TABLE VI

Incorporation of radiosulfate into various algal sulfur compounds during the process of dark-greening<sup>a</sup>

	Time (hr)								
		0	3	9	15	24	48	72	
Protein and polysaccharide	$\frac{1}{2}$	14 55	$\begin{array}{c} 144 \\ 604 \end{array}$	1,540 7,700	7,200 37,400	15,700 83,200	17,200 103,000	15,500 93,000	
SO4=	$\frac{1}{2}$	49 192	498 2,080	2,210 11,100	6,580 34,200	5,830 30,900	7,110 42,400	5,520 33,100	
Glutathione	$\frac{1}{2}$	0 0	18 75	105 525	380 1,980	588 3,120	960 5,720	917 5,500	
Sulfoquinovosyl glycerol	$1 \\ 2$	00	0 0	$\begin{array}{c} 41\\204\end{array}$	406 2,110	1,220 6,460	2,710 16,200	2,700 16,200	
Sulfolipid	$\frac{1}{2}$	0 0	0	75 376	297 1,540	800 4,240	728 4,340	657 3,940	
Lysosulfolipid	$\frac{1}{2}$	0 0	0	72 358	52 269	297 1,570	561 3,340	138 828	

<sup>a</sup> For explanation, see the legends of Table V.

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the total amount of sulfolipid are the resultant of its synthesis and breakdown. Marked net increase in the amounts of sulfate and sulfoquinovosyl glycerol were observed in the cells during the process of bleaching. The total <sup>35</sup>S-activity in the protein and polysaccharide fraction remained almost constant throughout the experiment, its content in cells (per packed cell volume) decreasing considerably with the increase in the cell mass. The cellular concentration of glutathione markedly decreased as the bleaching proceeded, while those of several other water-soluble sulfur compounds decreased only gradually.

5. Incorporation of  ${}^{35}S$  into the sulfolipid and other sulfur compounds by algal cells during the processes of greening and bleaching

The modes of incorporation of radiosulfate into the sulfolipid and other sulfur-containing compounds by algal cells were studied during the processes of their greening (in the light (Expt. A-II) and in darkness (Expt. A-III)) and bleaching (Expt. B-V). Tables V, VI and VII show the results

	Time (hr)									
		0	3	9	15	24	48	72		
Protein and polysaccharide	$\frac{1}{2}$	19 53	56 163	260 905	402 1,930	913 4,660	1,500 12,700	2,620 33,200		
SO <sub>4</sub> =	$1 \\ 2$	234 655	1,700 4,890	3,990 13,900	3,100 14,900	4,510 23,000	5,680 47,800	6,810 86,200		
Glutathione	$\frac{1}{2}$	0 0	0 0	36 126	$\begin{array}{c} 104 \\ 499 \end{array}$	$\begin{array}{c} 151 \\ 770 \end{array}$	$144 \\ 1,220$	189 2,390		
Sulfoquinovosyl glycerol	$1 \\ 2$	0 0	0 0	91 317	266 1,280	$1,170 \\ 5,970$	2,120 17,900	2,610 33,000		
Sulfolipid	$1 \\ 2$	0 0	28 82	$\begin{array}{c} 205 \\ 713 \end{array}$	392 1,880	769 3,920	481 4,060	496 6,280		
Lysosulfolipid	$rac{1}{2}$	0 0	0	13 46	43 205	77 392	$\begin{array}{c} 22\\181 \end{array}$	50 628		

TABLE VII

Incorporation of radiosulfate into various algal sulfur compounds during the process of bleaching<sup>a</sup>

<sup>a</sup> For explanation, see the legends of Table V.

of the experiments. During the greening and bleaching processes investigated the most striking mode of radiosulfate incorporation was observed for the sulfolipid. Figure 5 reproduces the time courses of <sup>35</sup>S-incorporation into the sulfolipid in the three different transition processes. In the control experiment shown in the figure, the non-<sup>35</sup>S-labeled green cells were incubated, in the dark, in the basal mineral medium containing radiosulfate (Expt. C-VII). The time courses of the <sup>35</sup>S-incorporation under these different conditions were found to be identical during the initial 24 hr period of

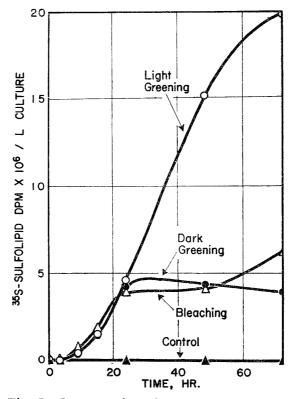


Fig. 5. Incorporation of <sup>35</sup>S into the sulfolipid in the four different transition processes of algal cells.

the experiments. During the following period of experiments, however, sharply different courses of the incorporation were observed: active incorporation continued throughout in the light-greening, while the incorporation abruptly ceased in the dark-greening and bleaching.

It has been revealed in the previous study (12) that the process of greening of the glucose-bleached cells consists of two consecutive phases: an early light-independent phase in which nucleic acids play an important role and the following light-dependent one in which full organization of chloroplasts takes place. It is during the latter light-requring phase that the formation of lamellar structures as well as active syntheses of chlorophyll and protein with the appearance of full photosynthetic activities takes place. In the dark, on the other hand, these light-requiring processes remain arrested. The results obtained here strongly suggest that the sulfolipid is an integral part of the photosynthetic structures, and that the biosynthesis of this lipid is closely related to the formation of the photosynthetic apparatus in chloroplasts. The observation that <sup>35</sup>S-incorporation into the sulfolipid came to a standstill at about the 24th hr in the bleaching experiment, may be taken as an indication that a repressive action of glucose on the sulfolipid synthesis started around that stage.

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#### TABLE VIII

Changes in radioactivity of the <sup>35</sup>S-labeled sulfur compounds of the green cells during the process of bleaching in a medium containing no radiosulfate ("chasing" experiment)<sup>a</sup>

		Time (hr)							
		0	3	9	15	24	48	72	
Protein and polysaccharide			14,000 39,400		$11,100 \\ 46,600$	9,740 49,700	6,170 48,800	4,830 48,100	
SO <sub>4</sub> =	$1 \\ 2$	1,880 5,280	2,180 6,120	$1,130 \\ 4,110$	$1,160 \\ 4,900$	$1,040 \\ 5,310$	898 7,100	1,140 11,000	
Glutathione	$\frac{1}{2}$	2,830 7,950	$1,570 \\ 4,520$	1,090 3,990	$1,060 \\ 4,480$	682 3,500	239 1,890	138 1,330	
Sulfoquinovosyl glycerol	$\frac{1}{2}$	3,030 8,500	3,780 11,000	$2,740 \\ 9,940$	$2,940 \\ 12,400$	3,500 17,900	2,900 22,900	2,440 23,500	
Sulfolipid	$\frac{1}{2}$	1,380 3,880			664 2,790	638 3,260	139 932	96 923	
Lysosulfolipid	$\frac{1}{2}$	56 156			26 111	29 147	8 64	7 71	

<sup>a</sup> The green cells uniformly labeled with <sup>35</sup>S were incubated in a non-radioactive bleaching medium. For further explanation, see the legends of Tables III and V.

6. Fates of the <sup>35</sup>S-labeled compounds of the green cells during the process of bleaching in a medium containing no radiosulfate ("chasing" experiment)

The "G"-cells uniformly labeled with <sup>35</sup>S were transferred to the "bleaching medium," on the one hand (Expt. B-VI), and to the basal medium containing no radiosulfate on the other (Expt. C-VIII). Table VIII shows the changes in radioactivity of the major sulfur-containing compounds of the cells in the bleaching process. In the control culture incubated in the basal medium, the radioactivities of all compounds studied did not change significantly. During the process of bleaching, however, the concentrations of various labeled compounds changed in different ways sug-Most striking gesting different metabolic turnover of the compounds. were the behaviors of sulfolipid and 6-sulfo- $\alpha$ -p-quinovopyranosyl- $(1 \rightarrow 1')$ -pglycerol. Figure 6 reproduces the characteristic changes in radioactivity of the sulfolipid and glycoside. The <sup>35</sup>S activity of the sulfolipid showed a slight decrease at an early stage but remained constant during the following period up to the 24th hr, after which time a marked decrease took place. This characteristic change may be interpreted as follews: (i) the degradation of the <sup>35</sup>S-sulfolipid proceeds continuously from the beginning of bleaching process except during the initial very short period, (ii) an active incorporation of <sup>35</sup>S from intracellular sources such as <sup>35</sup>SO<sub>4</sub><sup>=</sup> into the sulfolipid occurs during the initial period of about 24 hr, and (iii) the resultant of the degradation (loss of <sup>35</sup>S) of, and <sup>35</sup>S-incorporation into, the sulfolipid gives rise to the characteristic change in the <sup>35</sup>S-activity of the sulfolipid as observed above. The inference mentioned under (ii) is based on the observation described in the preceding section that an active incorporation of radiosulfate into the sulfolipid occurred during the initial 24 hr period of bleaching, becoming repressed afterwards. It seems highly likely that the degradation of the sulfolipid is a cardinal event in the bleaching process, proceeding in close association with the disappearance of chlorophyll and chloroplast structures.

The radioactivity of sulfoquinovosyl glycerol, unlike those of other sulfur-containing compounds, increased markedly during the bleaching process. It was surmised that the <sup>35</sup>S of the glycoside came from the <sup>35</sup>S-sulfolipid. Although the pathways of biosynthesis and metabolism of the sulfolipid have not yet been elucidated, the unique structure of this glycoside, the deacylated form of the sulfolipid, suggests its close metabolic relation to the sulfolipid<sup>3</sup>. Taking into consideration the fairly well established biosynthetic pathways of galactolipids (*18*), we are led to a plausible assumption that sulfoquinovosyl glycerol is not a biosynthetic precursor but

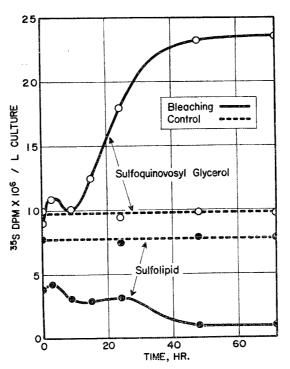


Fig. 6. Changes in radioactivity of the <sup>35</sup>Slabeled sulfolipid and sulfoquinovosyl glycerol during the process of cell bleaching in a medium containing no radiosulfate ("chasing" experiment).

<sup>&</sup>lt;sup>3</sup> The glycoside has never been found in organisms without photosynthetic activity so far investigated. Recently YAGI and BENSON demonstrated an enzymatic hydrolysis of the sulfolipid with formation of lysosulfolipid and sulfoquinovosyl glycerol (17).

the direct degradation product of the sulfolipid<sup>3</sup>. It is seen from the data in Fig. 6 or in Table VIII that during the process of bleaching the <sup>35</sup>S increment in the glycoside was much higher than the <sup>35</sup>S decrement in the sulfolipid. This result may appear to contradict with the above surmise. As mentioned earlier, however, the sulfolipid seems to be rapidly turning over during the initial 24 hr period of bleaching, incorporating <sup>35</sup>S from intracellular sources on the one hand, and being degraded, on the other, a situation that makes the apparently contradictory observation mentioned above understandable. In any event, it may be quite plausible that the <sup>35</sup>S-glycoside was, at least partly, a product of degradation of the <sup>35</sup>Ssulfolipid.

The deacylation of the sulfolipid, with formation of the water soluble glycoside, will naturally lead to a degradation of photosynthetic apparatus, if the sulfolipid is—by virtue of its surfactant nature—an architectural component in maintaining the functional structure of the apparatus, as has been suggested by BENSON (19). It may be added that the behavior of glutathione, which also showed a marked decrease in its <sup>35</sup>S-activity during the process of bleaching, is worthy of notice in connection with the processes of de- and re-generation of chloroplasts.

# CONCLUDING REMARKS

The present study has demonstrated that in *Chlorella protothecoides* the de- and re-generation of photosynthetic apparatus are causally related, respectively, to the degradation and synthesis of the sulfolipid discovered by BENSON et al. in various photosynthetic organisms. The important role of the sulfolipid in the maintenance of photosynthetic micro-structures, as has been suggested in the present study, seems to be also evidenced by our recent finding (SHIBUYA and MARUO, submitted to publication) that the sulfolipid and the lysosulfolipid are specifically concentrated in the quanta-some, a particulate fraction obtained from chloroplast (20).

During the course of this study, ROSENBERG and PECKER (21) reported similar results with *Euglena* cells in their process of light-induced greening.

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#### REFERENCES

 M. CALVIN, R. C. FULLER, G. MILHAUD and A. A. BENSON. Unpublished data; quoted in A. A. BENSON. 1961. Chloroplast lipid metabolism. *Vth International Congress of Biochemistry*. Moscow, August. Symposium No. VI, preprint No. 66.

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- (2) R. WISER. 1958. The radiochemical identification of the major sulfolipid in plants. *Ph. D. Thesis, The Pennsylvania State Univ.*
- (3) A. A. BENSON, H. DANIEL and R. WISER. 1959. A sulfolipid in plants. Proc. Natl. Acad. Sci., U. S., 45, 1582-1587.
- (4) A. A. BENSON and I. SHIBUYA. 1961. Surfactant lipids. In *Physiology and Biochemistry of Algae*. Edited by R. A. LEWIN. p. 371-383. Academic Press, New York, N. Y.
- (5) H. DANIEL, M. MIYANO, R. O. MUMMA, T. YAGI, M. LEPAGE, I. SHIBUYA and A. A. BENSON. 1961. The plant sulfolipid. Identification of 6-sulfo-quinovose. J. Am. Chem. Soc., 83, 1765-1766.
- (6) M. LEPAGE, H. DANIEL and A. A. BENSON. 1961. The plant sulfolipid. II. Isolation and properties of sulfoglycosyl glycerol. *ibid.*, 83, 157-159.
- (7) I. SHIBUYA and A. A. BENSON. 1961. Hydrolysis of  $\alpha$ -sulfo-quinovosides by  $\beta$ -galactosidase. Nature, 192, 1186-1187.
- (8) M. MIYANO and A. A. BENSON. 1962. The plant sulfolipid. VI. Configuration of the glycerol moiety. J. Am. Chem. Soc., 84, 57-59.
- (9) M. MIYANO and A. A. BENSON. 1962. The plant sulfolipid. VII. Synthesis of 6-sulfo-α-D-quinovopyranosyl-(1→1')-glycerol and radiochemical syntheses of sulfolipids. *ibid.*, 84, 59-62.
- (10) A. A. BENSON. 1963. plant sulfolipid. Advances in Lipid Research., 1, 387-394.
- (11) I. SHIHIRA-ISHIKAWA and E. HASE. 1964. Nutritional control of cell pigmentation in *Chlorella protothecoides* with special reference to the degeneration of chloroplast induced by glucose. *Plant & Cell Physicl.*, 5, 227-240.
- (12) S. AOKI and E. HASE. 1964. De- and re-generation of chloroplasts in the cells of *Chlorella protothecoides*. I. Syntheses of nucleic acids and protein in relation to the process of regeneration of chloroplast. *ibid.*, 5, 473-484.
- (13) D. I. ARNON. 1938. Microelements in culture-solution experiments with higher plants. Am. J. Bot., 25, 322-325.
- (14) K. SHIBATA. 1959. Spectrometry of translucent biological materials—Opal glass transmission method. In *Methods of Biochemical Analysis.* 7. Edited by D. GLICK. p. 77-109. Interscience Publishers, Inc., New York, N.Y.
- (15) A. A. BENSON, J. A. BASSHAM, M. CALVIN, T. C. GOODALE, V. A. HAAS and W. STEPKA. 1950. The path of carbon in photosynthesis. V. Paper chromatography and radioautography of the products. J. Am. Chem. Soc., 72., 1710-1718.
- (16) I. SHIBUYA, T. YAGI and A. A. BENSON. 1963. Sulfonic acids in algae. In Studies on Microalgae and Photosynthetic Bacteria. Edited by Japanese Soc. Plant Physiologists. p. 627-636. Univ. Tokyo Press, Tokyo.
- (17) T. YAGI and A. A. BENSON. 1962. Plant sulfolipid. V. Lysosulfolipid formation. Biochim. Biophys. Acta, 57. 601-603.
- (18) E. F. NEUFELD and C. W. HALL. 1964. Formation of galactolipids by chloroplasts. Biochem. Biophys. Res. Comm., 14, 503-508.
- (19) A. A. BENSON. 1964. Plant membrane lipids. In Annual Review of Plant Physiology. 15. Edited by L. MACHLIS and W. R. BRIGGS. p. 1-16. Annual Reviews. Inc., Palo Alto, Calif.
- (20) R. B. PARK and N. G. PON. 1961. Correlation of structure with function in Spinacea oleracea chloroplast. J. Mol. Biol., 3, 1-10.
- (21) A. ROSENBERG and M. PECKER. 1964. Lipid alterations in Euglena gracilis cells during light-induced greening. Biochemistry, 3, 254-258.